





Introduction

Cost-Effective Valorization of Factory Tea Waste: Enhancing Circular Economy Practices

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Results and discussions



FTIR, XRD, TGA and Zeta potential

The optimal polyphenol extraction conditions were a water-tofactory tea waste ratio of 30:1 mL g⁻¹, temperature of 80°C, and extraction time of 60 min. At the optimized condition, the recorded polyphenol content was 2635.36 ± 52.07 mg GAE/L.

Characterization of the extract

IR spectroscopy indicates only minor alterations in the OH, CH, and C=O functional groups, yet substantial changes are observed in the NO2 and CN stretches, suggesting modifications in the interactions of nitro and nitrile groups. XRD analysis of both HC and F-HC samples demonstrates the presence of similar amorphous carbon phases, with distinctions in crystallinity and peak intensities that imply variations in crystallite sizes and lattice strain. Furthermore, zeta potential measurements reveal values of -30.3 mV for HC and -24.5 mV for activated HC, indicating a shift in surface charge that could impact the interaction dynamics with surrounding molecules.

Dye adsorption

Methodology

extraction

The adsorption study indicates that all observed adsorption behavior adheres to the Langmuir adsorption isotherm, suggesting a monolayer adsorption process with uniform energy distribution across the adsorbent surface.

Summary

Optimal Extraction: Efficiently extracted polyphenols from tea waste to maximize yield. Hydrochar Production: Converted residue to hydrochar, enhanced via Fenton oxidation. **Dye Adsorption:** Employed activated hydrochar for effective dye adsorption. Sustainability Impact: Transformed tea waste into valuable products, promoting environmental sustainability.

References

Debnath, B., Haldar, D., & Purkait, M. K. (2021). Potential and sustainable utilization of tea waste: A review on present status and future trends. Journal of Environmental Chemical Engineering, 9(5), 106179.

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