

# Acid modified MXenes for catalytic PET depolymerization

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## Motivation & Objectives

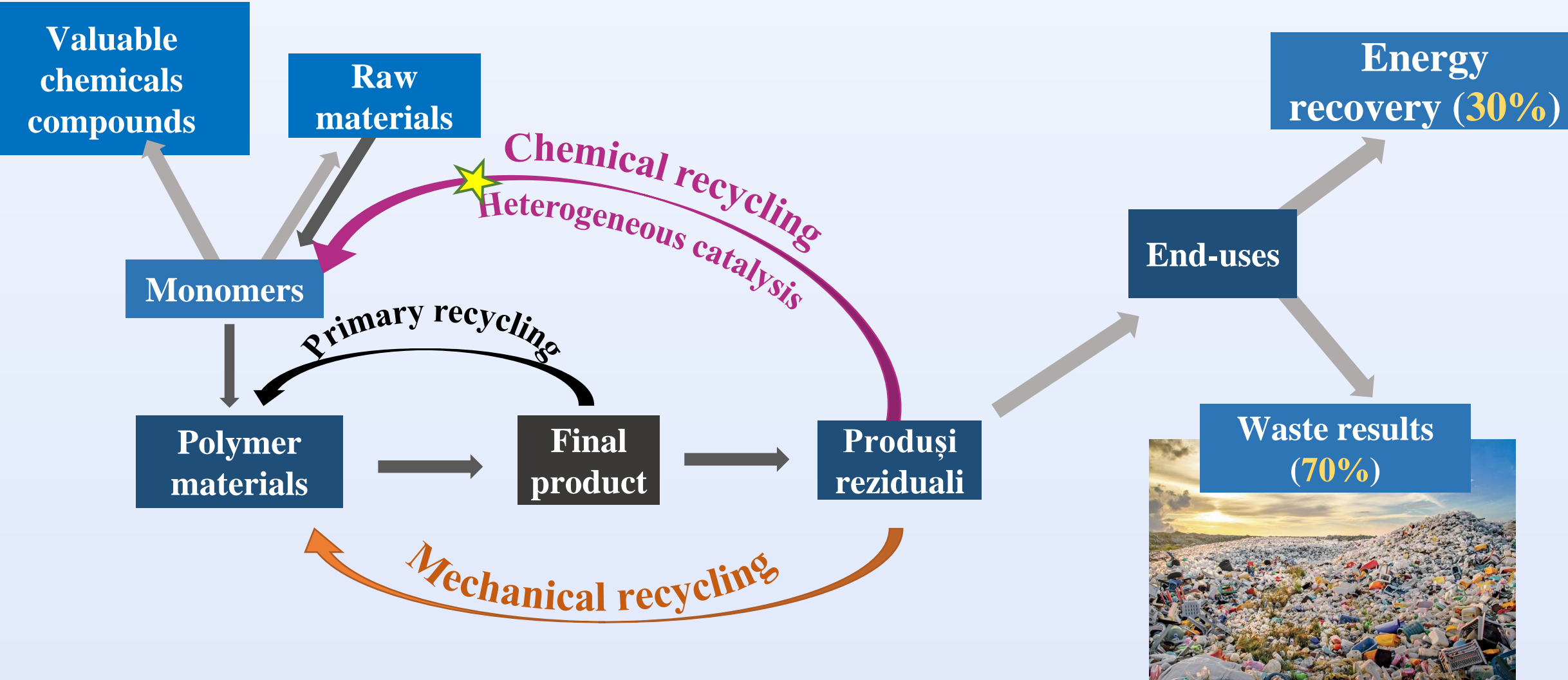
Plastics - a challenge for society? How?

High consumption

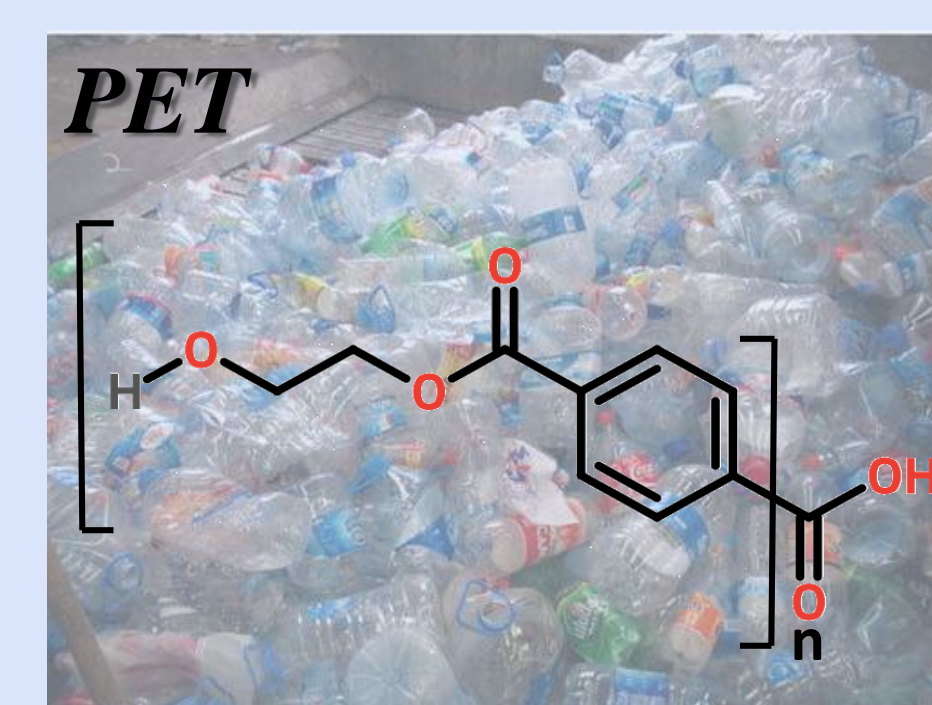
Large quantities of plastic waste

Bad plastic waste management!

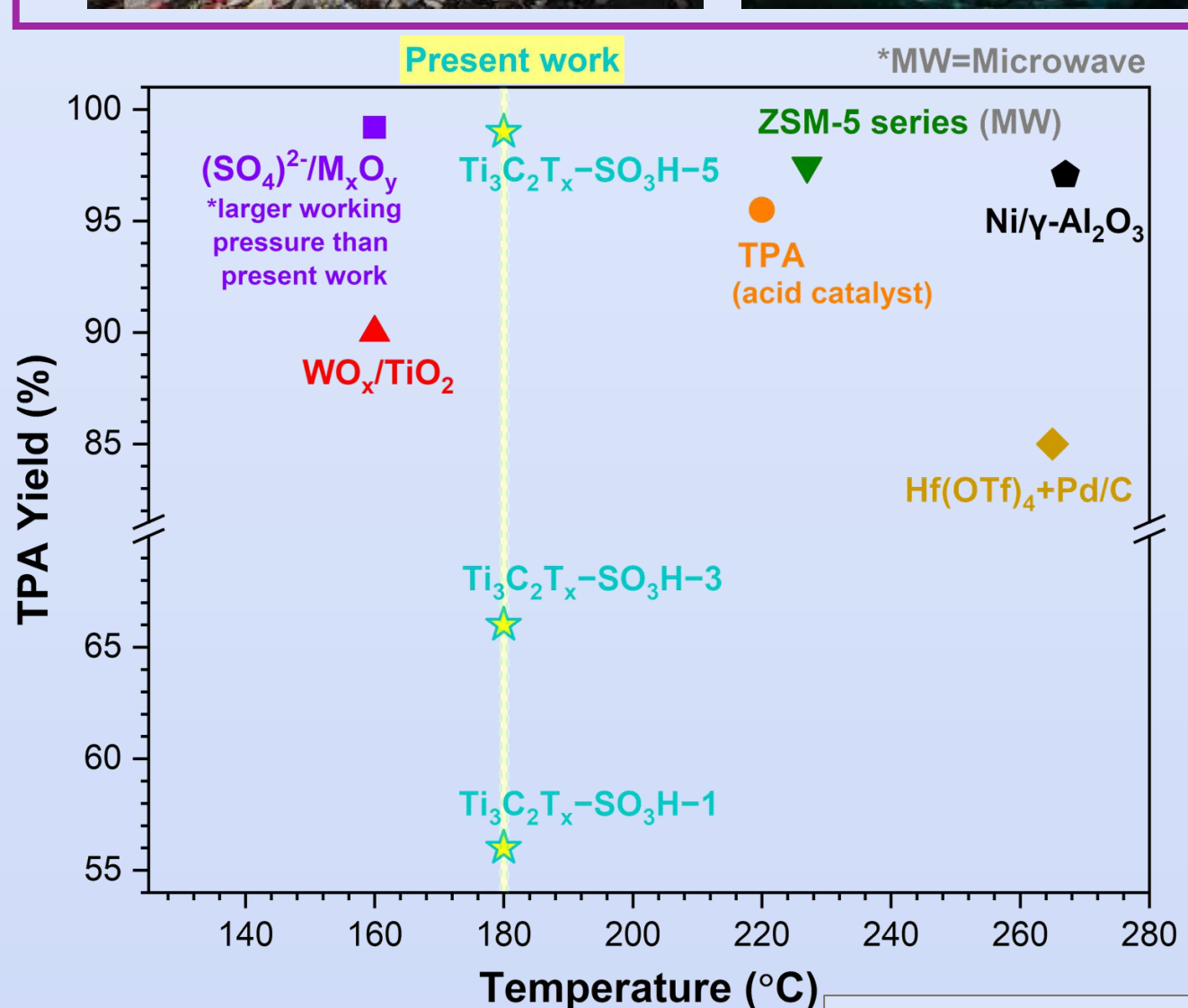
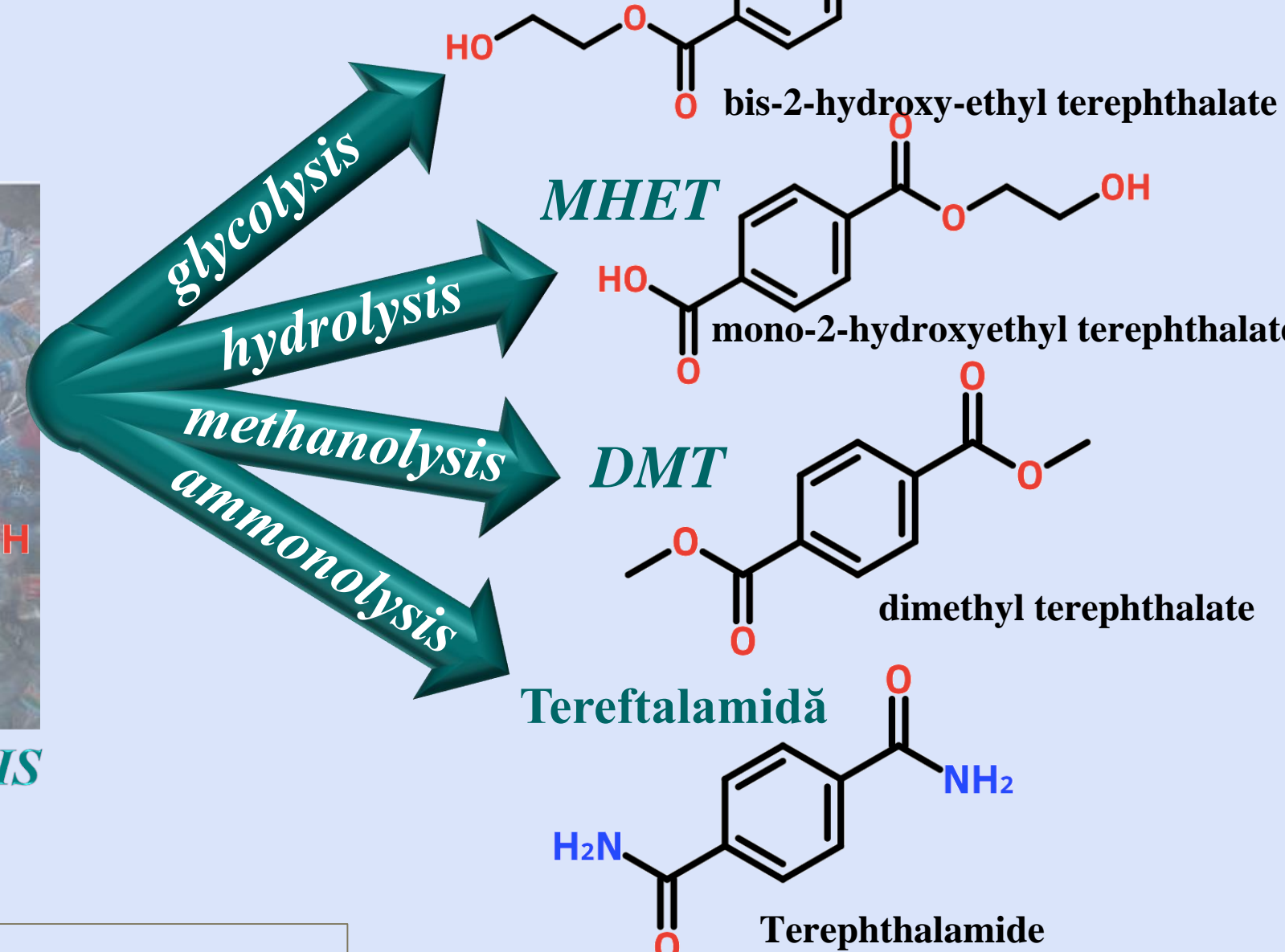
Crisis situation



## PET chemical recycling processes



HETEROGENEOUS CATALYSIS



## Context of PET depolymerization

Depolymerization is an acid-catalysed process → Strong acid sites

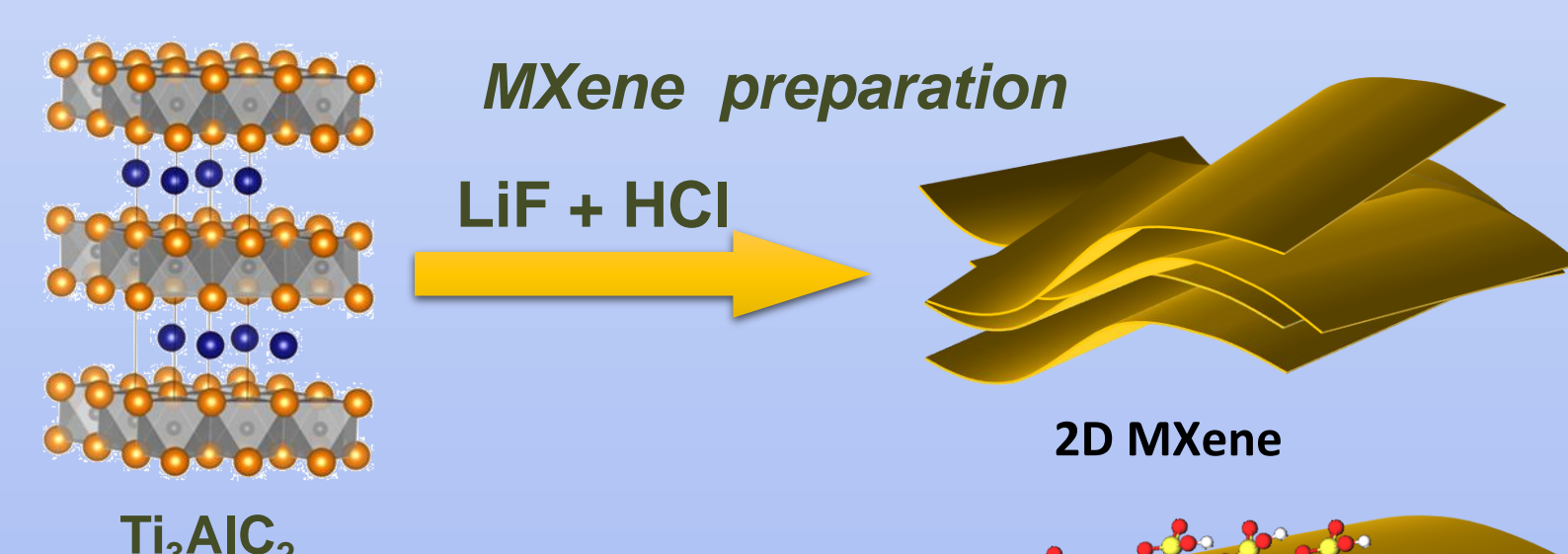
LIMITATIONS of heterogeneous catalytic systems:

– medium-high acidity → temperature

– microporous structure → accessibility

**SOLUTION:** a material with strong, stable and accessible acid centers to enable depolymerization at lower temperatures and higher reaction rates.

Catalyst preparation

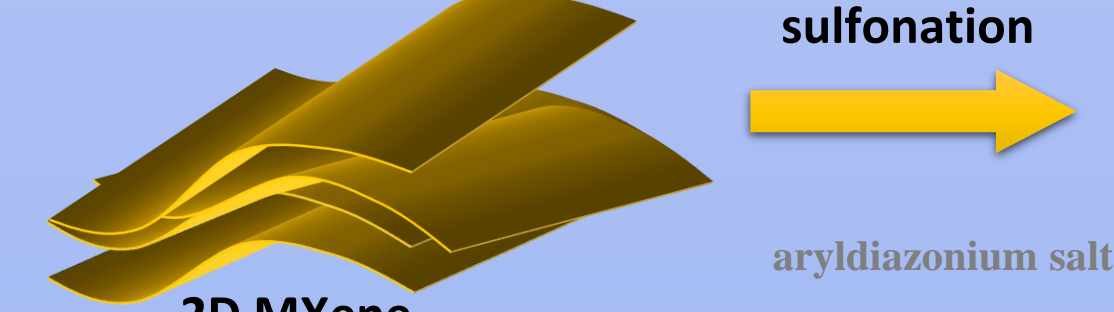


MXene preparation

LiF + HCl

2D MXene

Modification of MXene

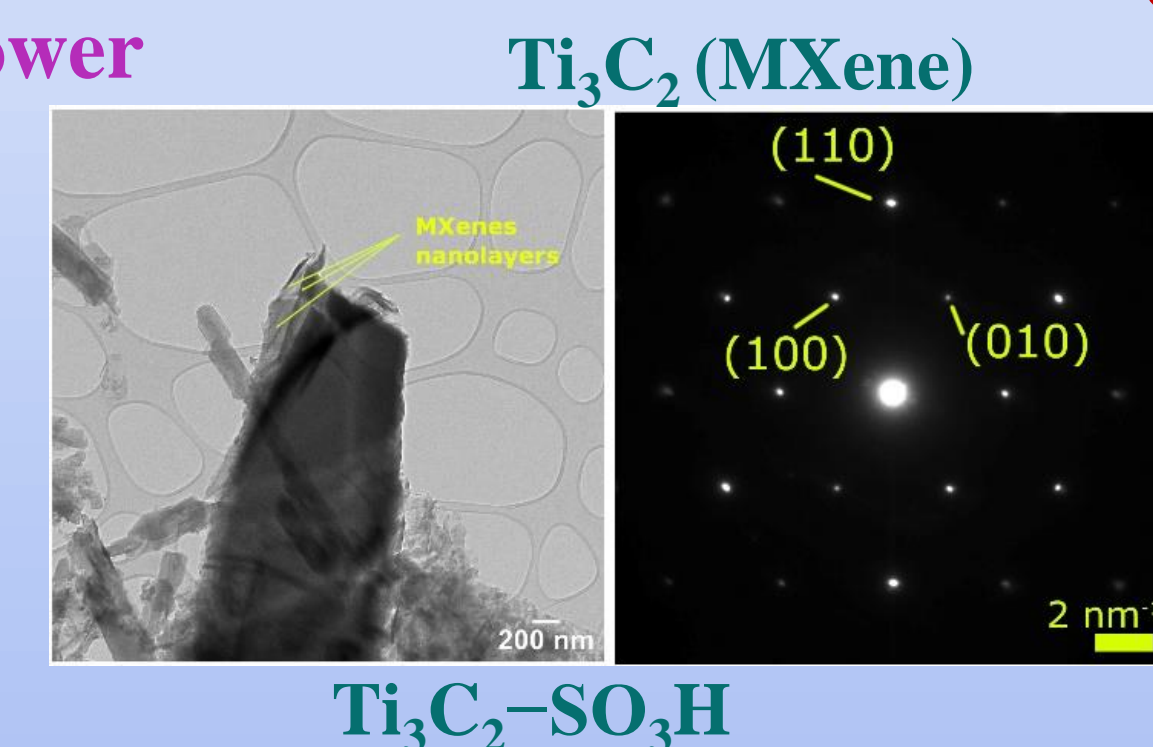


sulfonation

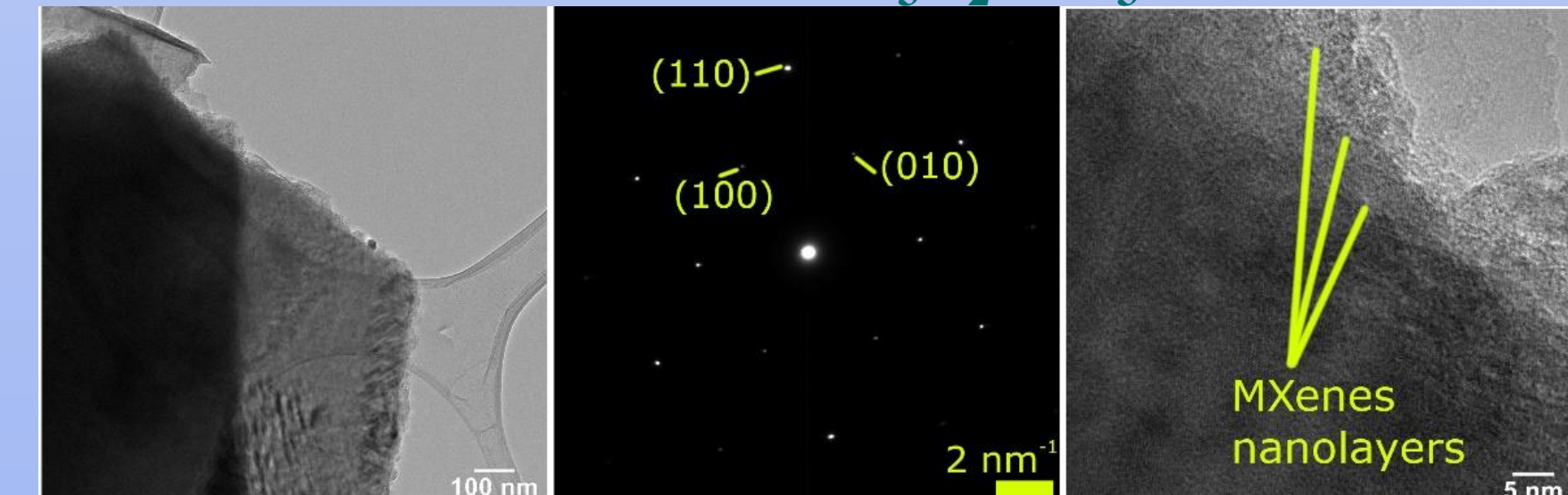
aryldiazonium salt

sulfonated 2D MXene

CATALYST	Ti (%)	C (%)	O (%)	N (%)	S (%)
SEM	TEM	SEM	TEM	SEM	TEM
Ti <sub>3</sub> AlC <sub>2</sub>	40.8	40.2	4.0	-	-
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub>	67.4	36.0	12.1	48.8	15.2
Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> -SO <sub>3</sub> H	51.7	45.0	16.6	44.2	26.7



Ti<sub>3</sub>C<sub>2</sub>-SO<sub>3</sub>H



Ti<sub>3</sub>C<sub>2</sub>-SO<sub>3</sub>H

## Catalytic reaction – Experimental part:

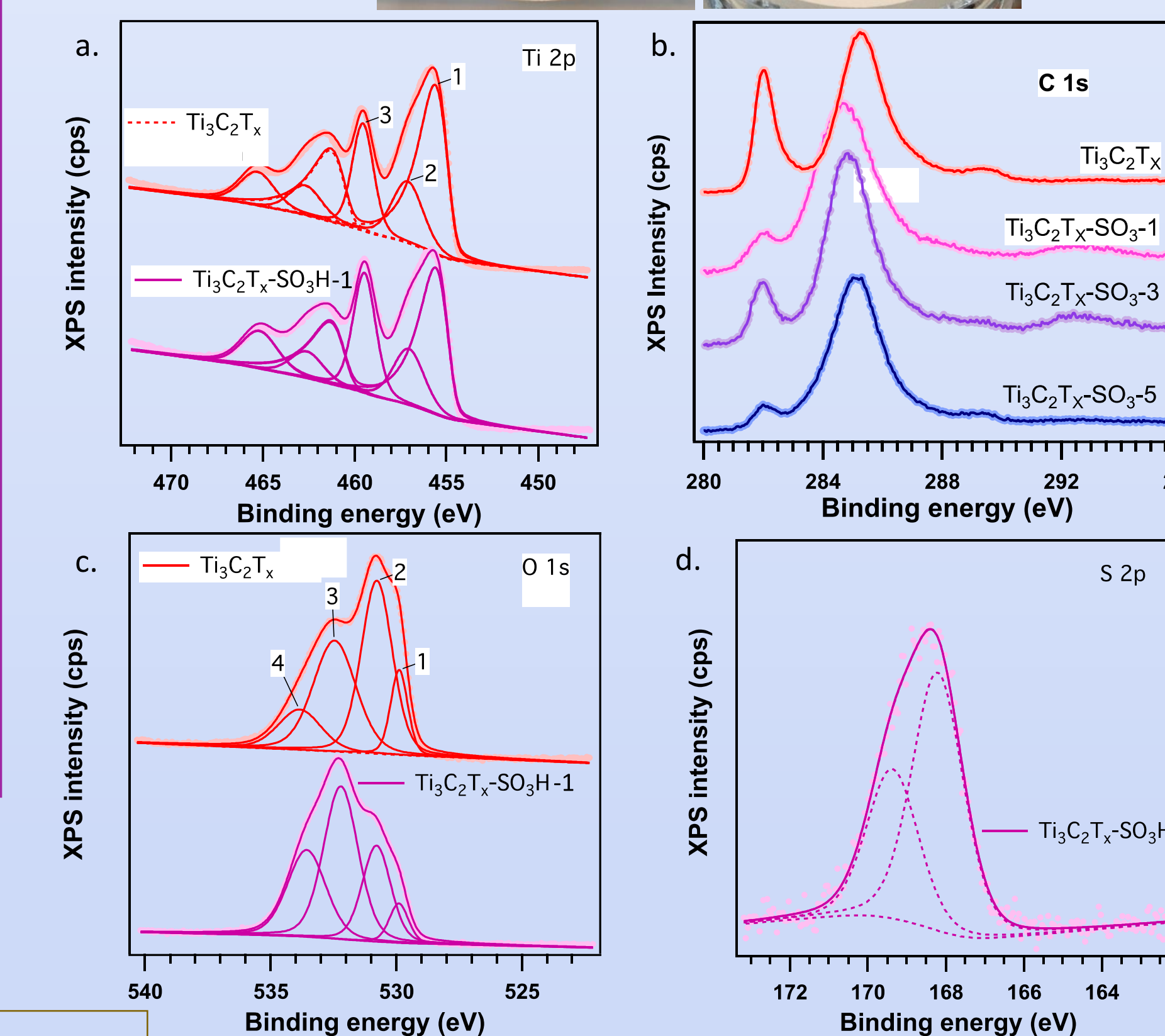
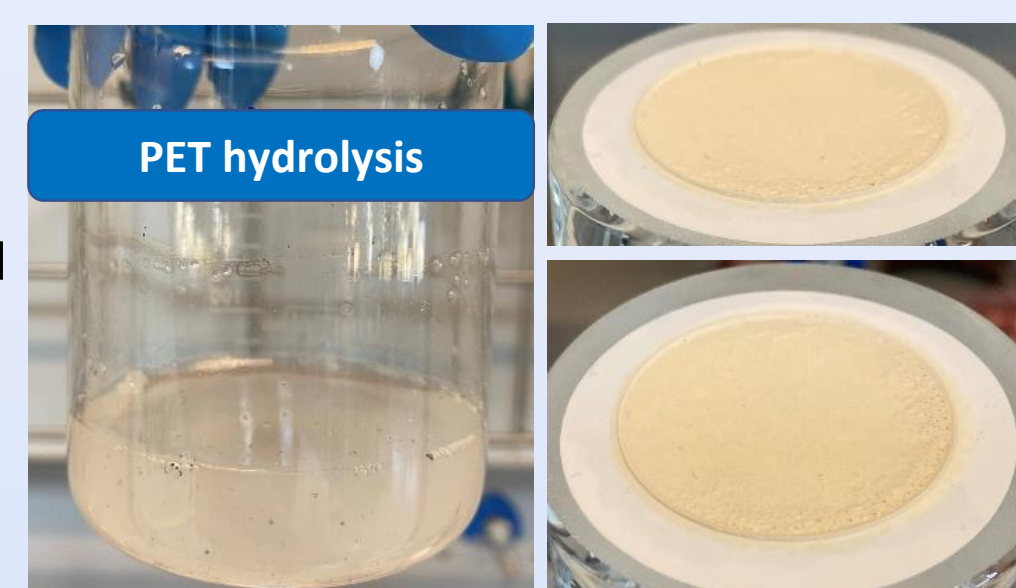
### Reaction conditions

- ✓ Reaction temperature (160-180 °C)
- ✓ Reaction pressure (10 bars N<sub>2</sub>)
- ✓ PET (100 mg) –chips 3x2 mm
- ✓ Catalyst (20-60 mg)
- ✓ Reaction time (24h)
- ✓ Solvent: H<sub>2</sub>O

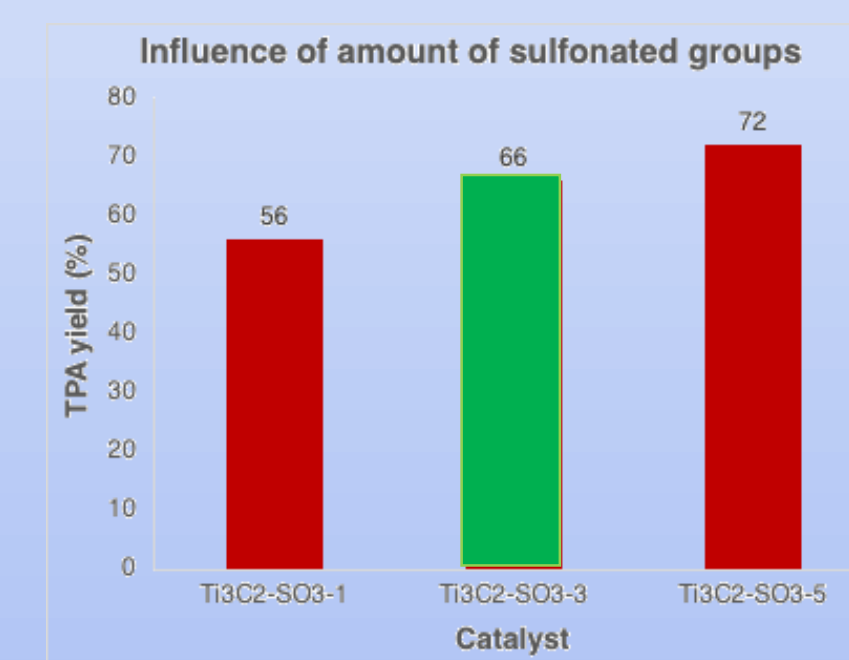
\*TPA = terephthalic acid

### Post-treatment

- ✓ Addition of 1M NaOH solution to form TPA-Na<sup>+</sup>
- ✓ Filtration
- ✓ Precipitation of TPA\* with HCl

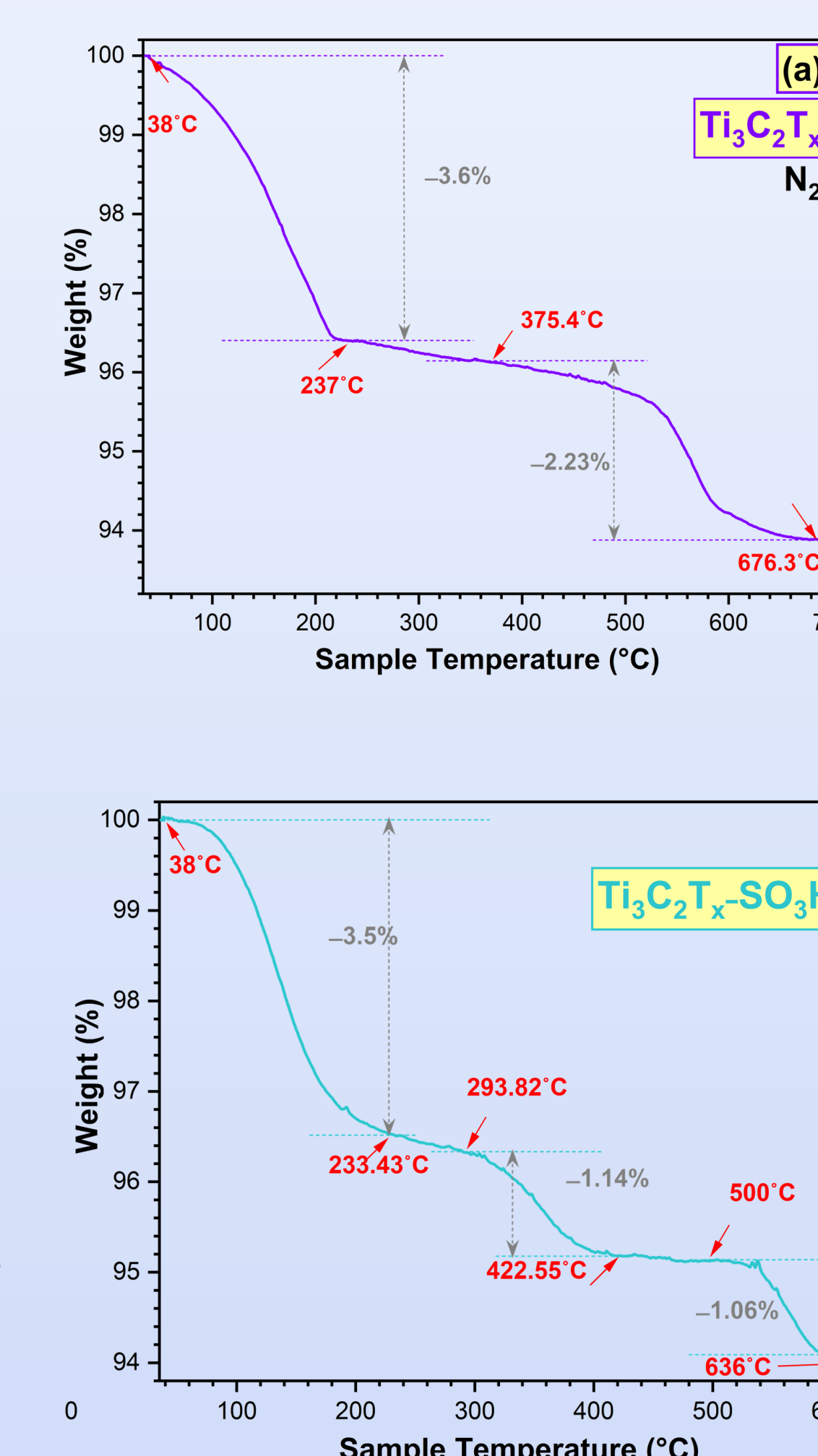


## Catalytic reaction - Results

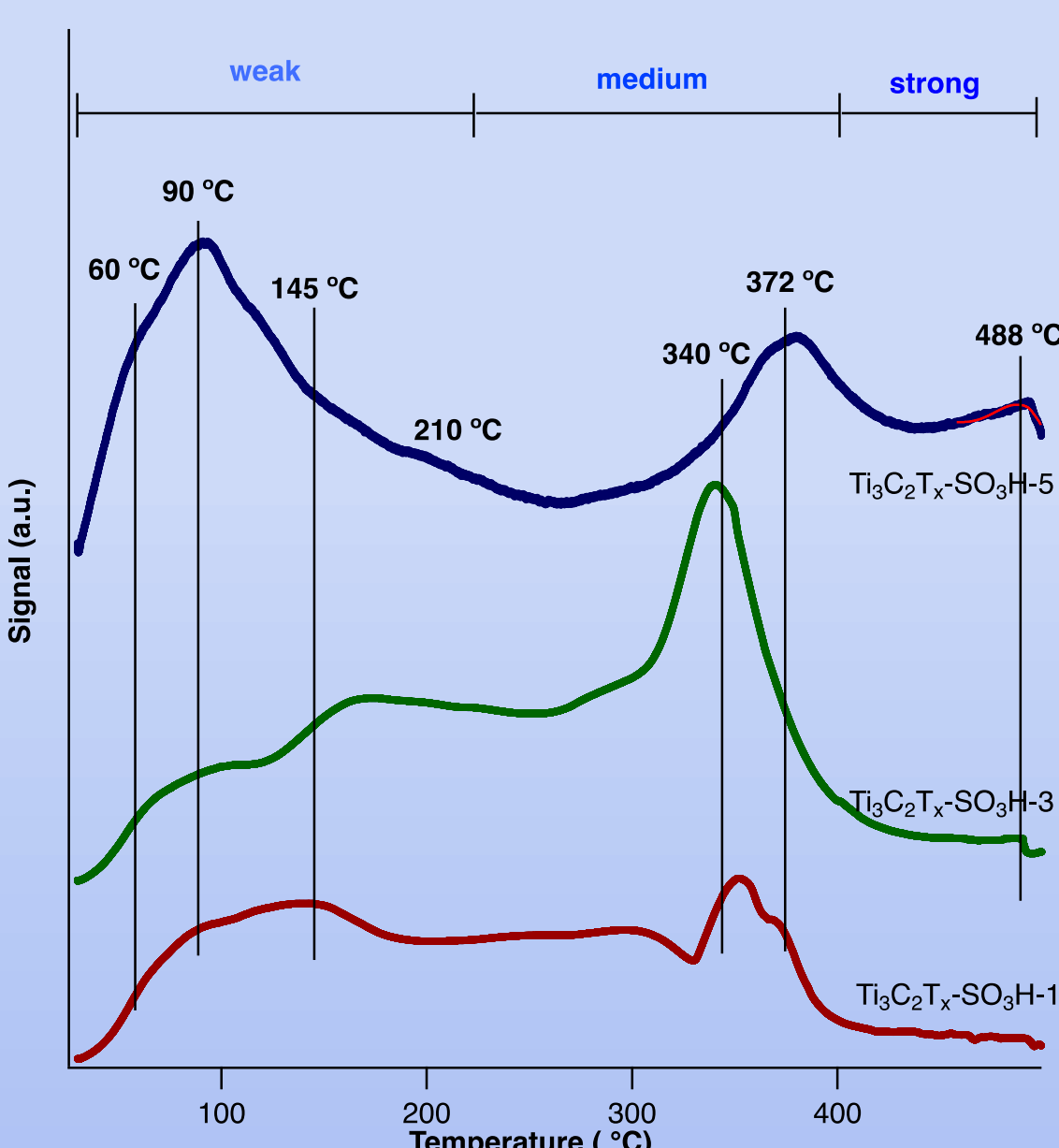


180 °C; 40 mg catalyst

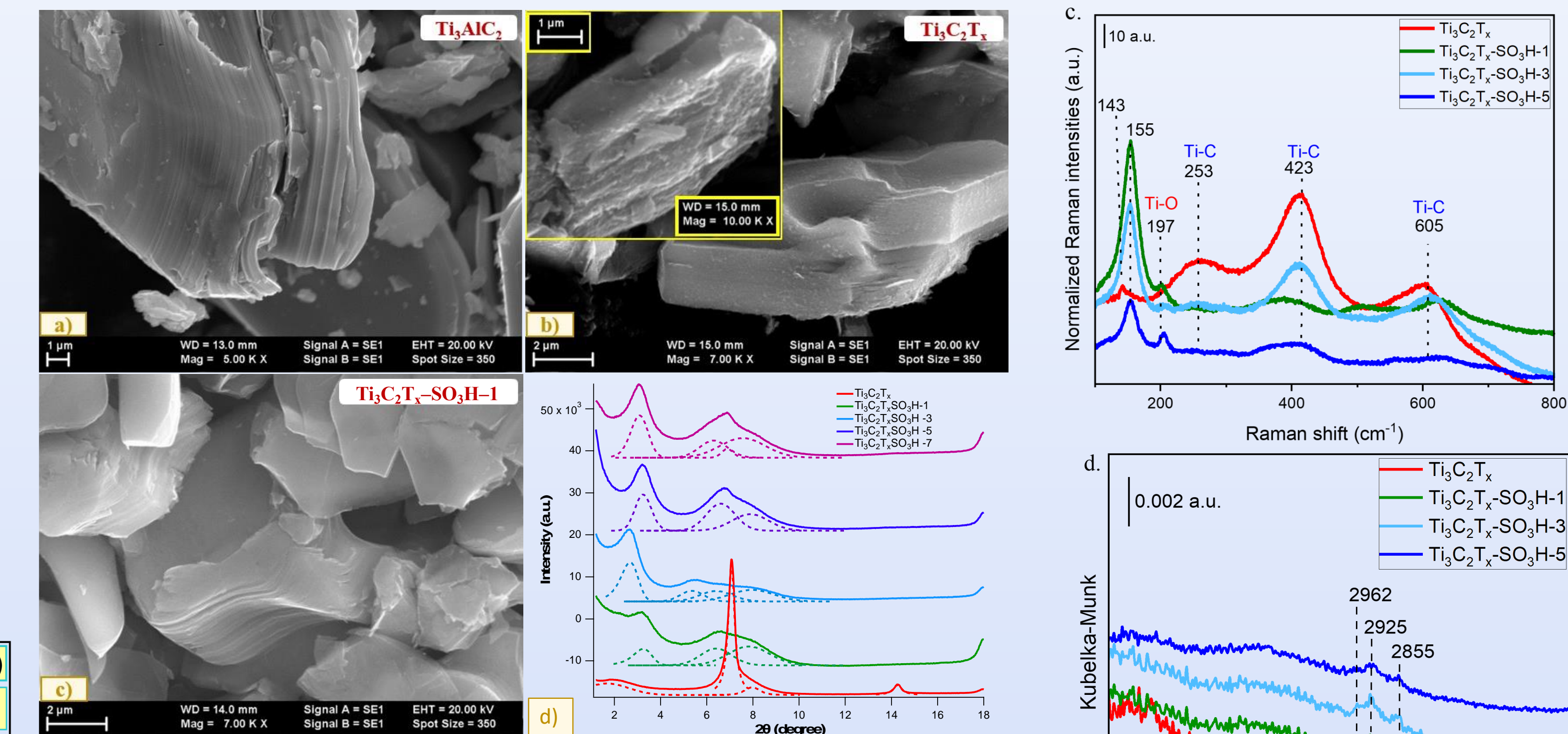
ACIDITY	Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> -SO <sub>3</sub> -1	Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> -SO <sub>3</sub> -3	Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> -SO <sub>3</sub> -5
(mmol/g)			
weak	964	734	2841
medium	644	1452	704
strong	-	-	77
Total	1608	2186	3622



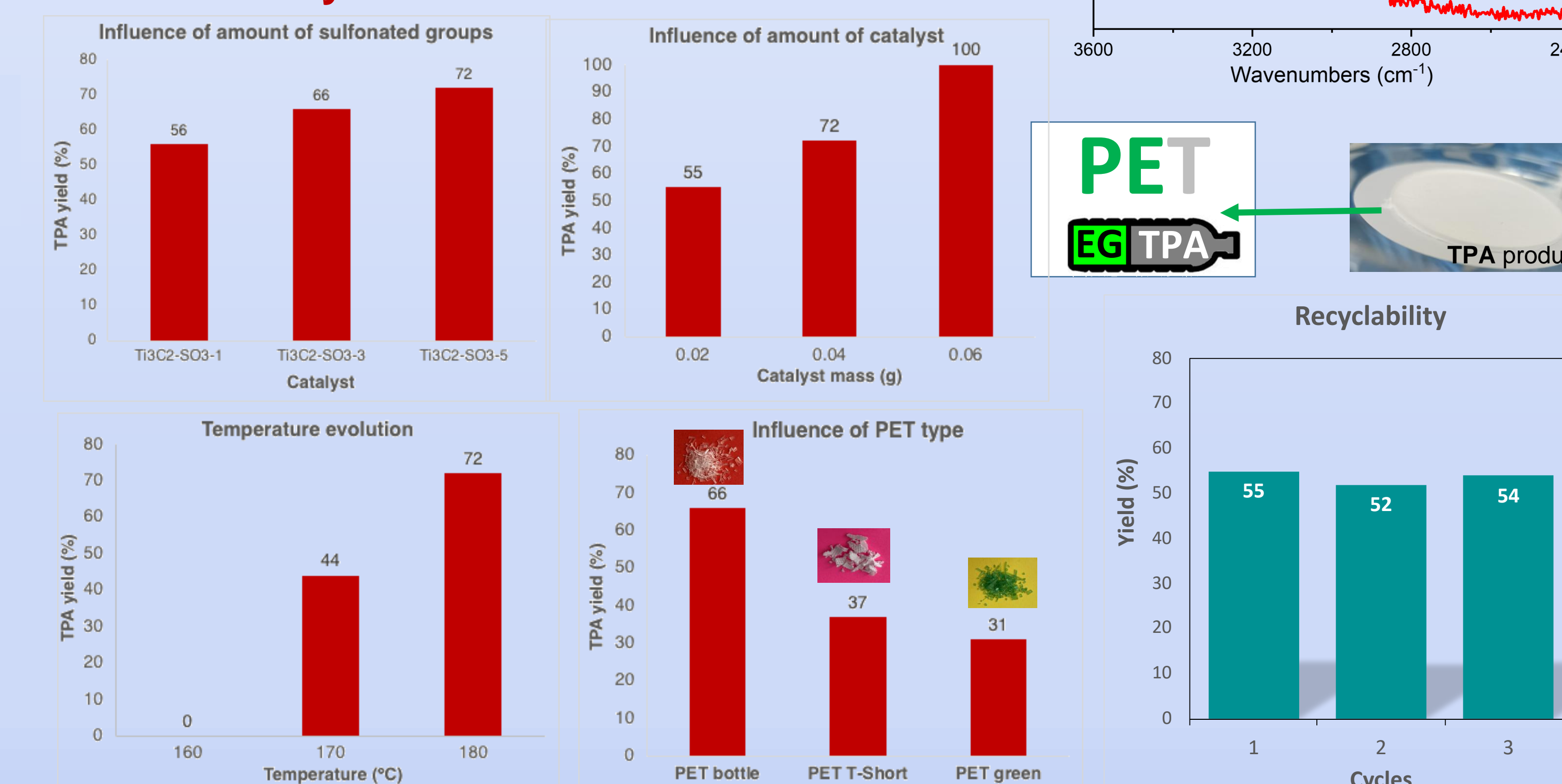
## Catalytic reaction - Results



## Results & Discussion



## Catalytic reaction - Results



## Conclusions

- ✓ Using aryl diazonium salt is a good approach to modify MXene with –SO<sub>3</sub>H groups on its surface.
- ✓ These materials are efficient in the PET depolymerization.
- ✓ The amount of sulfonic groups influence the yield of TPA.
- ✓ A higher temperature leads to a higher TPA yield.
- ✓ These materials are efficient for different types of PET. However, additives in PET influence negatively the depolymerization.